II. Enzymes

Catalysis	the rate of chemical reactions	
being used up in the pre	ocess	
Enzymes are	that act as biological catalysts	
	the activation energy of a reaction	
Activation Energy =		
	D	
†	A ↓ B	
Free Energy -	c	
	-	
	Time	
Enzymes have		(G)
Enzymes have of the reaction, they just	on the	(G)
of the reaction, they jus	on the st force the reaction to occur	
of the reaction, they jus	on the	
of the reaction, they just Reactants can only brea	on the st force the reaction to occur akdown when they have absorbed enough energy to	
of the reaction, they jus	on the st force the reaction to occur akdown when they have absorbed enough energy to	

- Parts of an Enzyme
 - Substrate =
 - Active Site =
- When the enzyme binds to the substrate, a temporary

Enzymes emerge from the reaction in their ______.

Induced Fit Model = refers to a ______ in shape to produce a ______ around the substrate

How do enzymes lower the activation energy?

_	Allows reactants to get into the
_	to react Stretches the substrate to causeeasier
_	(helps reach the transition state faster)
_	Creates a more conducive for the
	reaction to occur
_	participation of the active site in the reaction – may bind to the substrate temporarily
C. The active	site is an enzyme's catalytic center
oo s	ubstrate Products O
	Enzyme Enzyme-substrate
· ·	complex
D. A cell's ph	ysical and chemical environment affects enzyme activity
	ting Enzyme Activity
	Up to a certain point, the velocity of an enzymatic reaction with temperatures
•	Increased Temperature = increase in
	resulting in greater collisions between enzymes and molecules
•	Temperatures that get too high result in thermal agitation of the enzyme molecule - This enzyme becomes
Graph	
2.	
•	The optimal range for most enzymes is pH 6-8
•	One exception: Pepsin: digestive enzyme in the stomach – works best at pH 2
•	An environment that is or
Graph	can an enzyme
2	
5. <u> </u>	Most enzymes can't tolerate environmental conditions that are
•	This will also cause the enzyme to
	·

4. Cofactors = activity		helpers that many enzymes require for catalytic	
acı	a. Inorganib. Organic	c cofactors include: Mg, Ca, Fe, Cu, Zn, & Mn cofactors are called	— (Co A) are also
		ATP, Vitamins, and	(CoA) are also
	ii. (coenzymes Cofactors can bind tightly to the active site of the or they may bind loosely and release along with the	
5. Enz	yme Inhibitoi	rs	
•	Irreversible		
		e inhibitor attaches to the enzyme by	
		ls, inhibition is usually irreversible	
•	Reversible I		• , , , •
		oitor attaches to the enzyme usingersible inhibition can be competitive or non-comp	
A.	Competitive	e – the normal substrate mo	lecule and
		for admission into the active site	
	They	y reduce the productivity of enzymes	
	Incre	easing the	
	redu	ces the effectiveness of competitive inhibitors	
В.	-	itive – Impede the enzymatic reactions by bindin	g to a part of the
	- This	causes the enzyme to change its	rendering the active
	site i	unreceptive	
III. The Regu A. Metabolic		etabolism n depends on allosteric regulation	
Allosteric regi	ulation = invo	olves the	to an
		at its active site	
• This ca	auses the enz	yme to be or	
		yme to be or or depending on the molecule	
• ATP c	an inhibit act	ivity while ADP can stimulate activity on the sar	ne enzyme
Cooperativity	=	of enzyme activit	y when one substrate
binds to	active si	of enzyme activit ite and stimulates the	
Feedback inhi	bition = whe	n a metabolic pathway is	by the
inhibitory bind	ding of the	to an enzyme	<u> </u>
• This _	_	to an enzyme the pathway from continuing	
 Examp 	ole: synthesis	of amino acids	
D (D) 1 (1	e		

B. The location of enzymes within a cell helps order metabolism

- Some are grouped into complexes, some incorporated into membranes and others are contained inside organelles (lysosomes)

 Bacteria (prokaryotic cells) have enzymes located in the cytosol