

Ch 9: Cellular Respiration

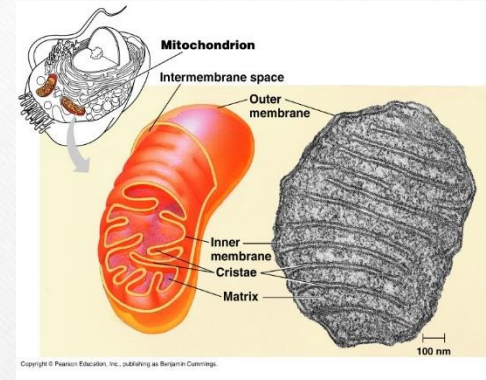


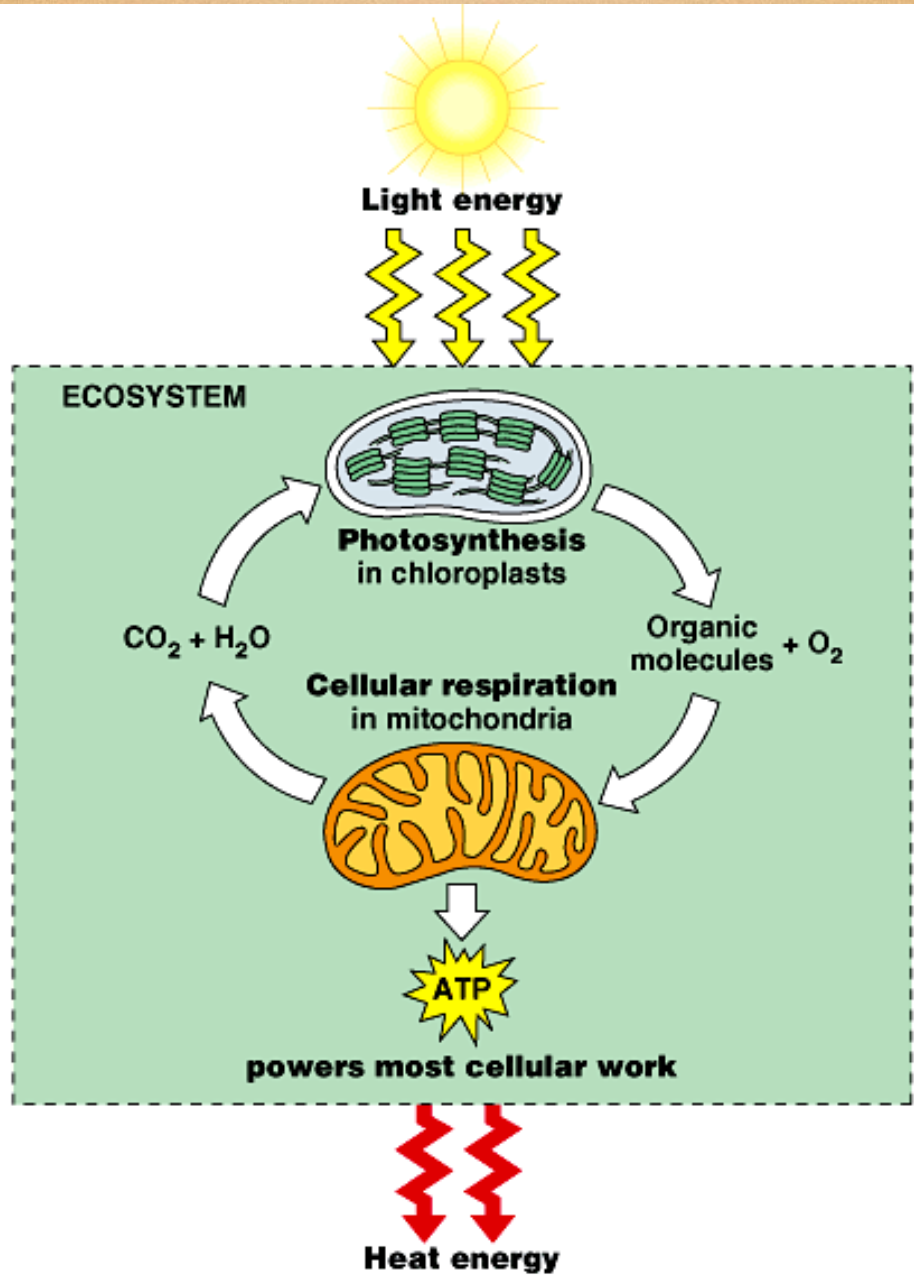
Cellular Respiration – An overview

- Exergonic reactions and catabolic pathway
- Energy stored in bonds of food molecules is transferred to ATP
- Cellular respiration provides the energy to make ATP from ADP and Phosphate
- Carbs, proteins, & lipids can be metabolized as fuel in cellular respiration
- Cells must replenish its ATP to do cellular work
- Occurs in the mitochondria

Mitochondria

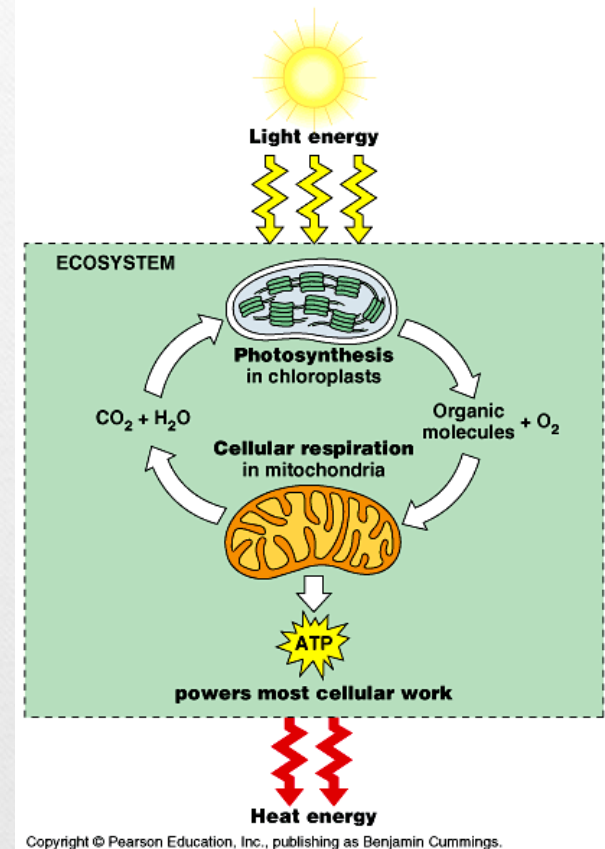
- Function: site of energy capture and transformation (cell respiration)
 - Quantity in cell correlated with metabolic activity
- Structure: double membrane (phospholipid bilayer)
 - Inner folds = Cristae; contain enzymes used in ATP production
 - Intermembrane space lies between the cristae and the outer membrane and the matrix makes up the middle of the mitochondrion
 - Contains its own DNA and can divide on its own





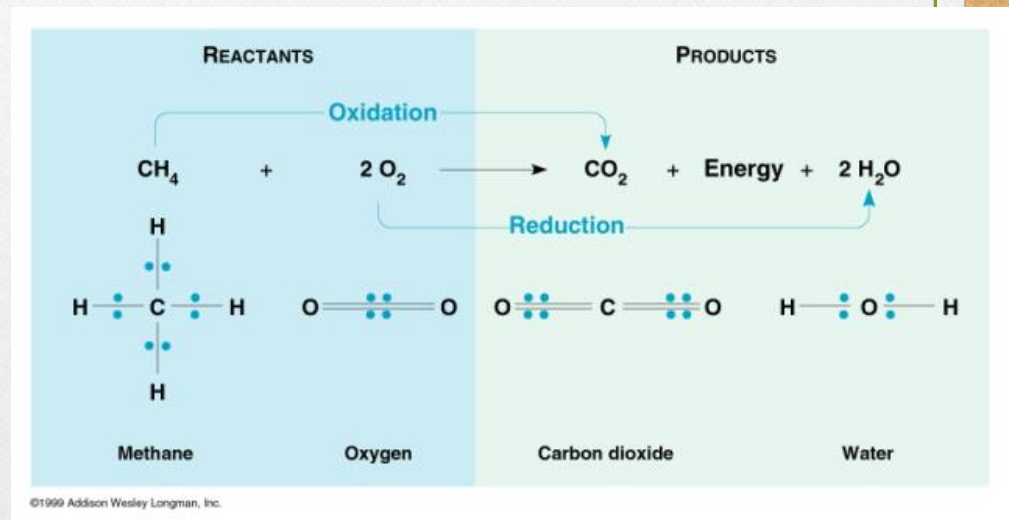
Principles of Energy Harvest

- Catabolic pathway
 - Cellular Respiration
 - $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O + \text{Energy (38 ATP + heat)}$
 - Can be from the breakdown of carbs, lipids, or proteins
 - Fermentation – occurs in the absence of O_2
 - Energy from these reactions are either used to do work or released as heat



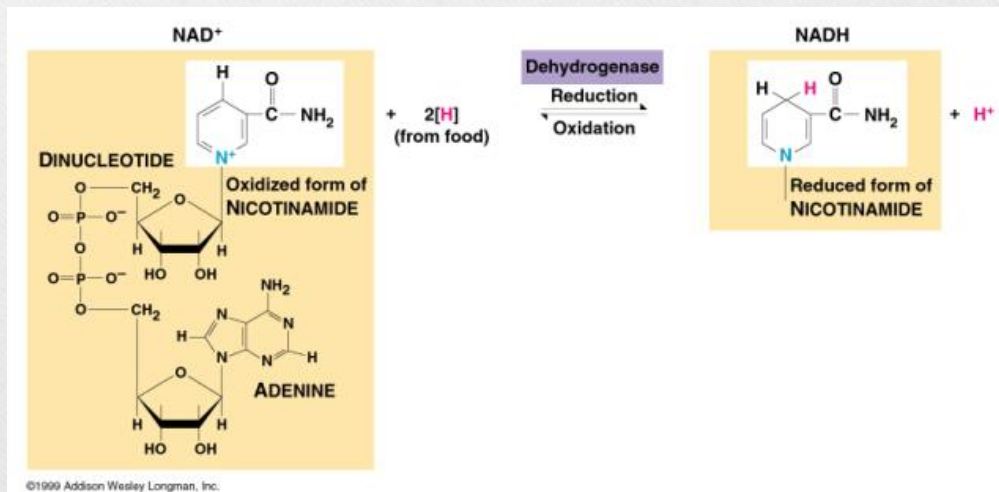
Oxidation-Reduction (Redox) Reactions

- Oxidation is e- loss;
reduction is e- gain
 - OIL RIG (adding e-
reduces + charge)
- Reducing agent is the
electron donor and
Oxidizing agent is
electron acceptor



Oxidizing Agent in Respiration

- NADH (nicotinamide adenine dinucleotide) and FADH_2 (flavin adenine dinucleotide)
 - NAD^+ is reduced to NADH, then oxidized to make ATP in the final step of cell respiration
 - Same process occurs with FADH_2



Pathway of Electrons

- Electrons are removed from food through a series of reactions to eventually get added to oxygen to make water
- Electron pathway: food \rightarrow NADH \rightarrow electron transport chain \rightarrow oxygen

Types of Phosphorylation

- Substrate-level Phosphorylation

- Type of phosphorylation in which a phosphate group is transferred from an intermediate compound to the recipient compound (ATP)

- Oxidative Phosphorylation

- Type of phosphorylation in which an inorganic phosphate group is added to ADP to form ATP through the oxidation of a molecule

Cellular Respiration – 2 Types

- Aerobic – O_2 is used in the production of ATP
 - Glycolysis → formation of Acetyl CoA → Krebs Cycle → ETC
- Anaerobic – no O_2 is present in the production of ATP; referred to as **FERMENTATION**
 - Glycolysis only

4 Steps of Aerobic Cellular Respiration

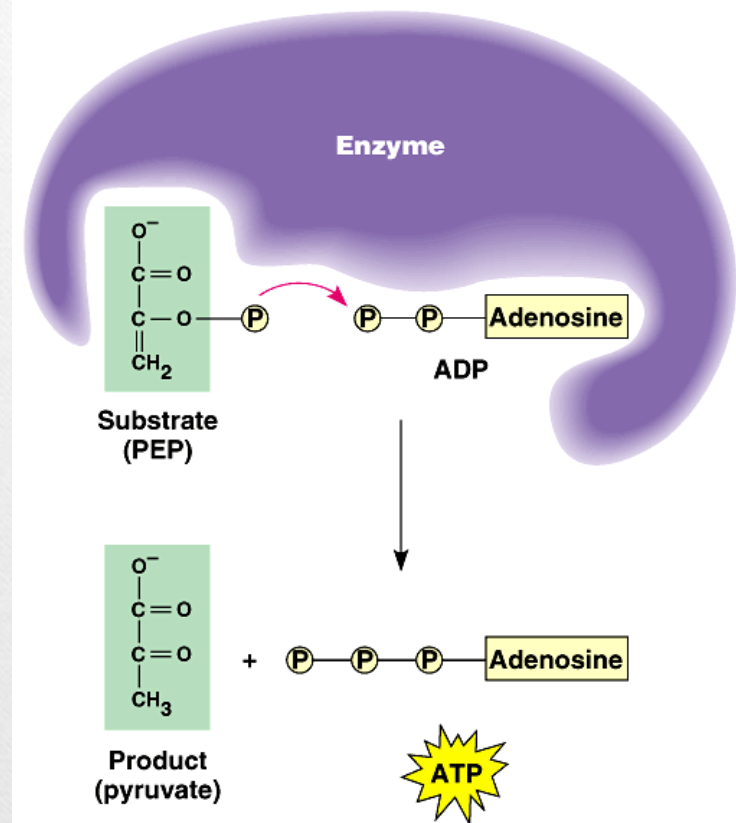
1. Glycolysis
2. Formation of acetyl CoA
3. Krebs Cycle
4. Electron Transport Chain (ETC)
with Chemiosmosis

STEP 1: Glycolysis = “splitting sugar”

- Glucose – a 6C sugar, is split to form 2 molecules of pyruvate – a 3C sugar
- Occurs in the cytosol
- Catabolic pathway that decomposes glucose and other organic fuels

Glycolysis

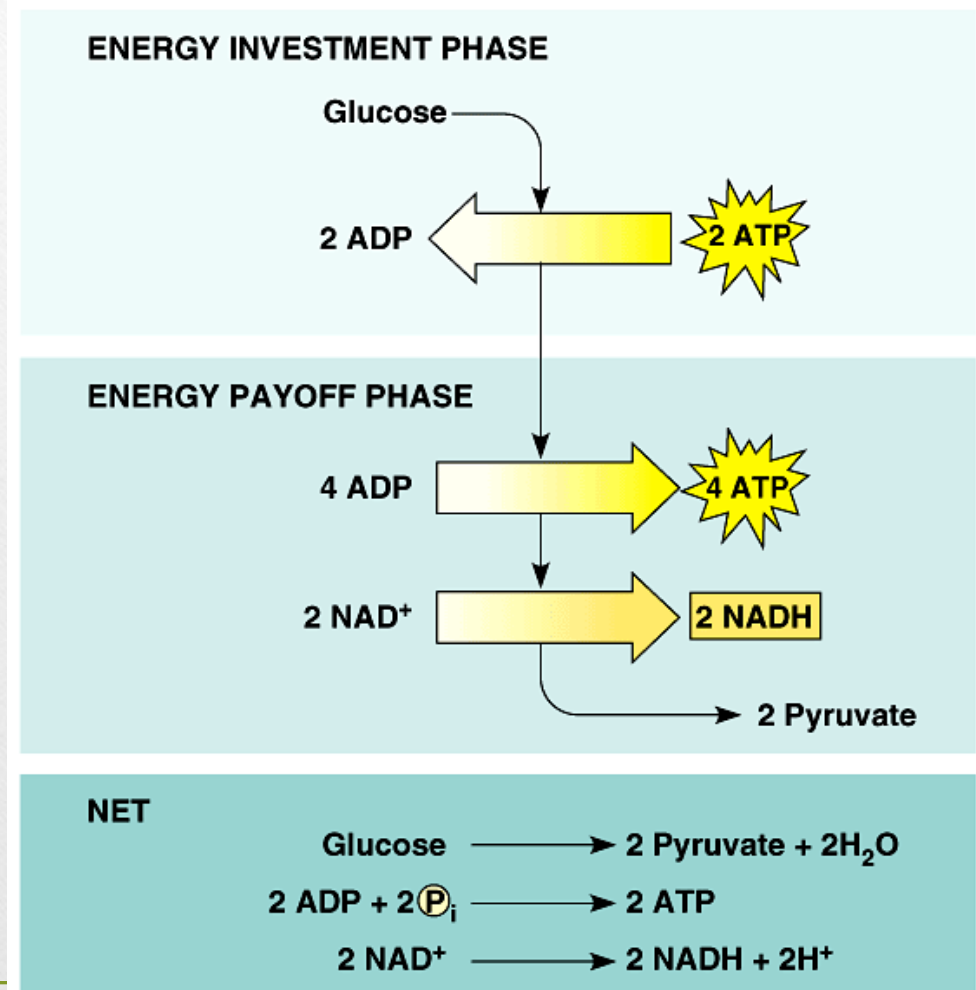
- 10 steps divided into 2 phases:
 - **Energy investment phase**
 - Cell uses 2 ATP in initial breakdown of glucose
 - **Energy payoff phase**
 - ATP is produced by substrate level phosphorylation and NAD⁺ is reduced to NADH by electrons released from the breakdown of food



Copyright © Pearson Education, Inc., publishing as Benjamin Cummings.

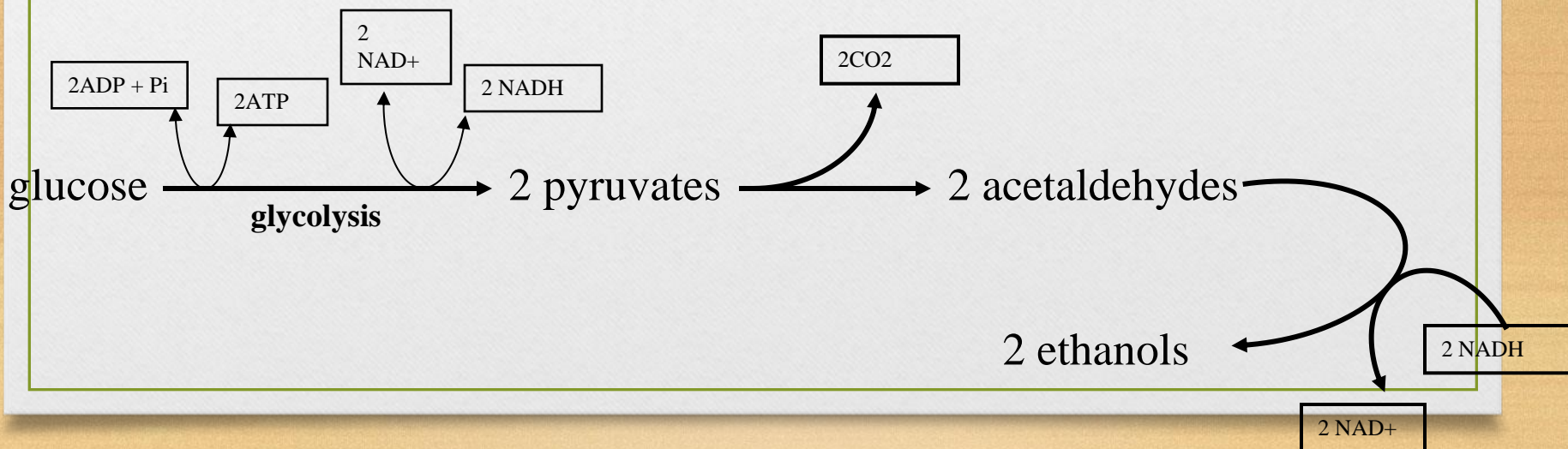
Net Energy Yield - Glycolysis

- 2 ATP produced
- 2 NADH produced



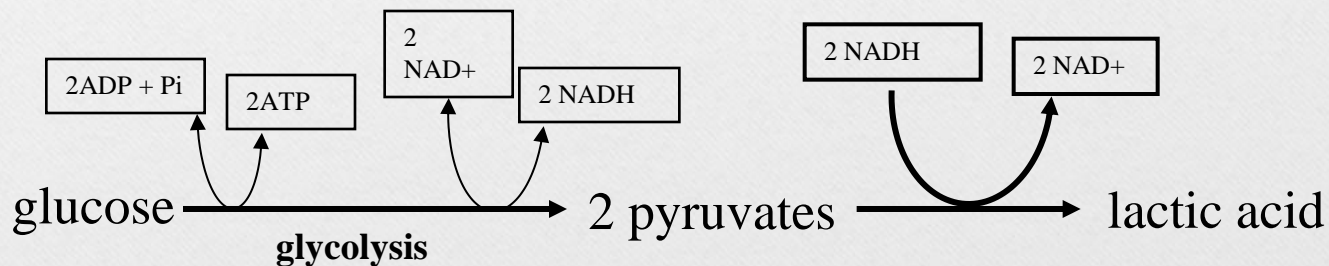
Types of Fermentation

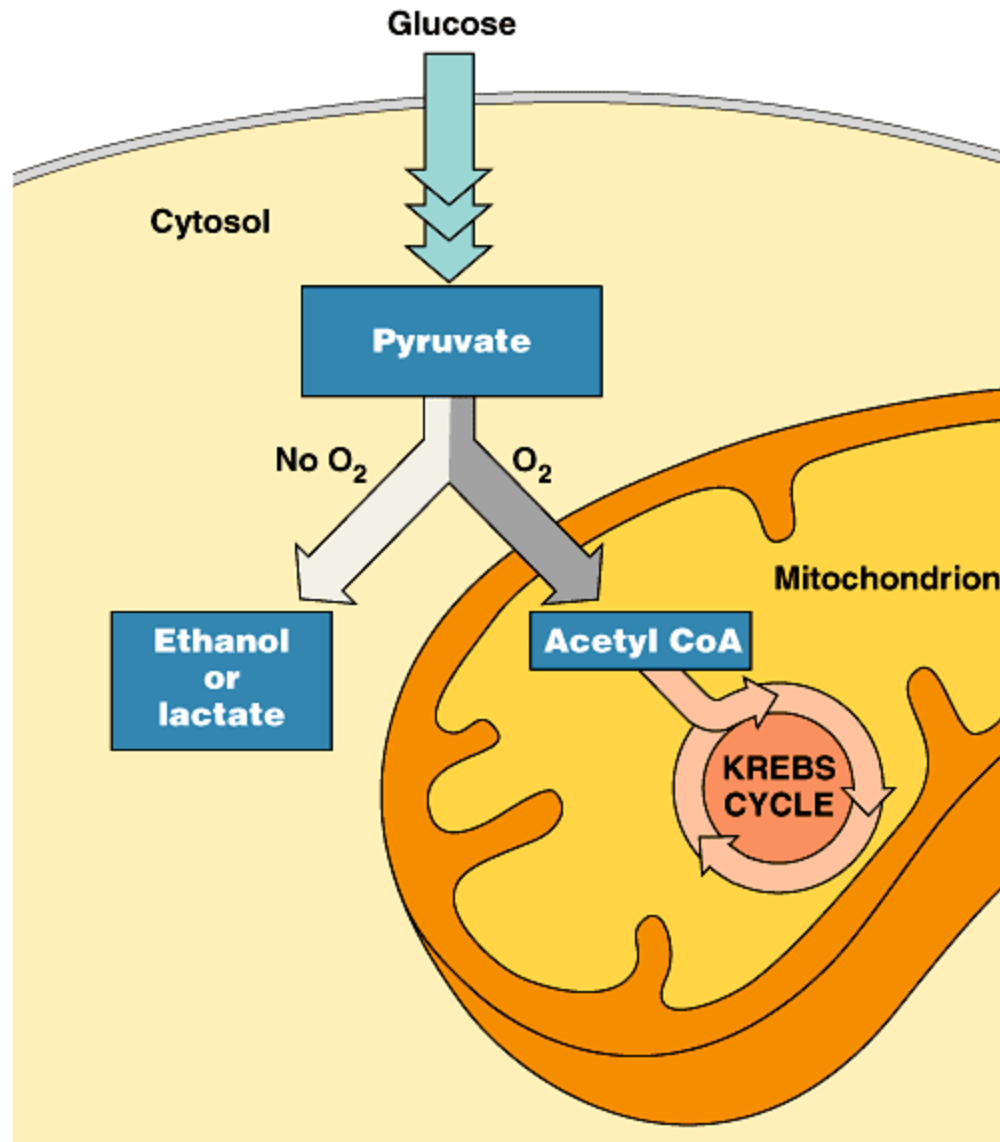
- **Alcohol Fermentation** – form of anaerobic respiration that converts glucose into ethanol and CO_2 by yeasts, fungi, or bacteria
 - Used in the production of alcoholic beverages and bread



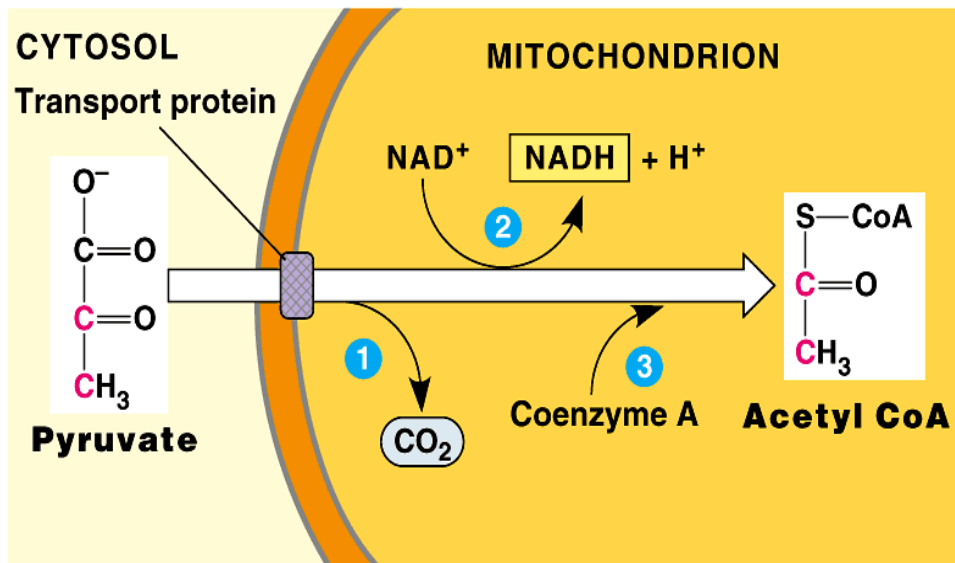
Types of Fermentation

- **Lactic Acid Fermentation**: form of anaerobic respiration that converts glucose into lactic acid by animals and some bacteria
 - Used in the dairy industry to make cheese & yogurt.
 - Human muscle cells create lactic acid & ATP





STEP 2 – Formation of Acetyl CoA



Copyright © Pearson Education, Inc., publishing as Benjamin Cummings.

- The junction between glycolysis and the Krebs's Cycle
- Pyruvate enters the mitochondria where it is oxidized into Acetyl CoA
 - Electrons are lost in the form of CO_2

Overall Reaction of Pyruvate \rightarrow acetyl CoA:



Steps of Acetyl CoA Formation

1. Pyruvate's carboxyl group (COO-) is removed and given off as a molecule of CO_2 .
2. Remaining 2C fragment is oxidized (electron is lost) forming a compound named acetate
 - An enzyme transfers the “lost” electrons to NAD+ forming NADH (equivalent to 3ATP molecules)
3. Coenzyme A, a sulfur-containing compound (derived from Vitamin B) is attached to the acetate by an unstable bond which makes the attached acetyl group highly reactive.

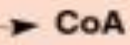
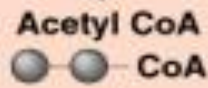
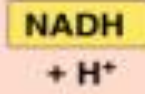
STEP 3 – Krebs Cycle

- A metabolic “furnace” that is also known as the Citric Acid Cycle
- Krebs Cycle reactions oxidize the remaining acetyl CoA into CO_2
- The Krebs Cycle is composed of 8 enzyme-controlled steps.
 - Two turns of the Krebs Cycle produces:
 - 2 ATP
 - 6 NADH (1 NADH is equal to 3 ATP)
 - 2 FADH_2 (1 FADH_2 1.5 ATP)

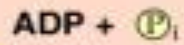
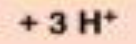
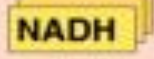
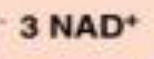
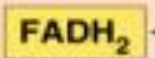
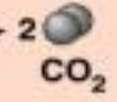
Krebs Cycle

- **Remember**: In Glycolysis, **one** molecule of glucose is partially oxidized into **2 molecules of pyruvate**. **Each** of the **2 pyruvate** molecules are then converted to Acetyl CoA which enters the Krebs cycle.

Pyruvate
(from glycolysis,
2 molecules per glucose)

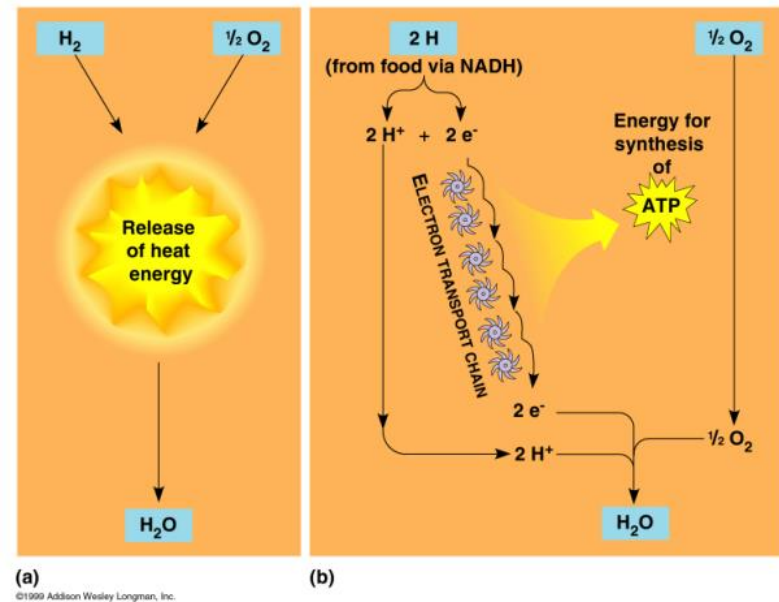


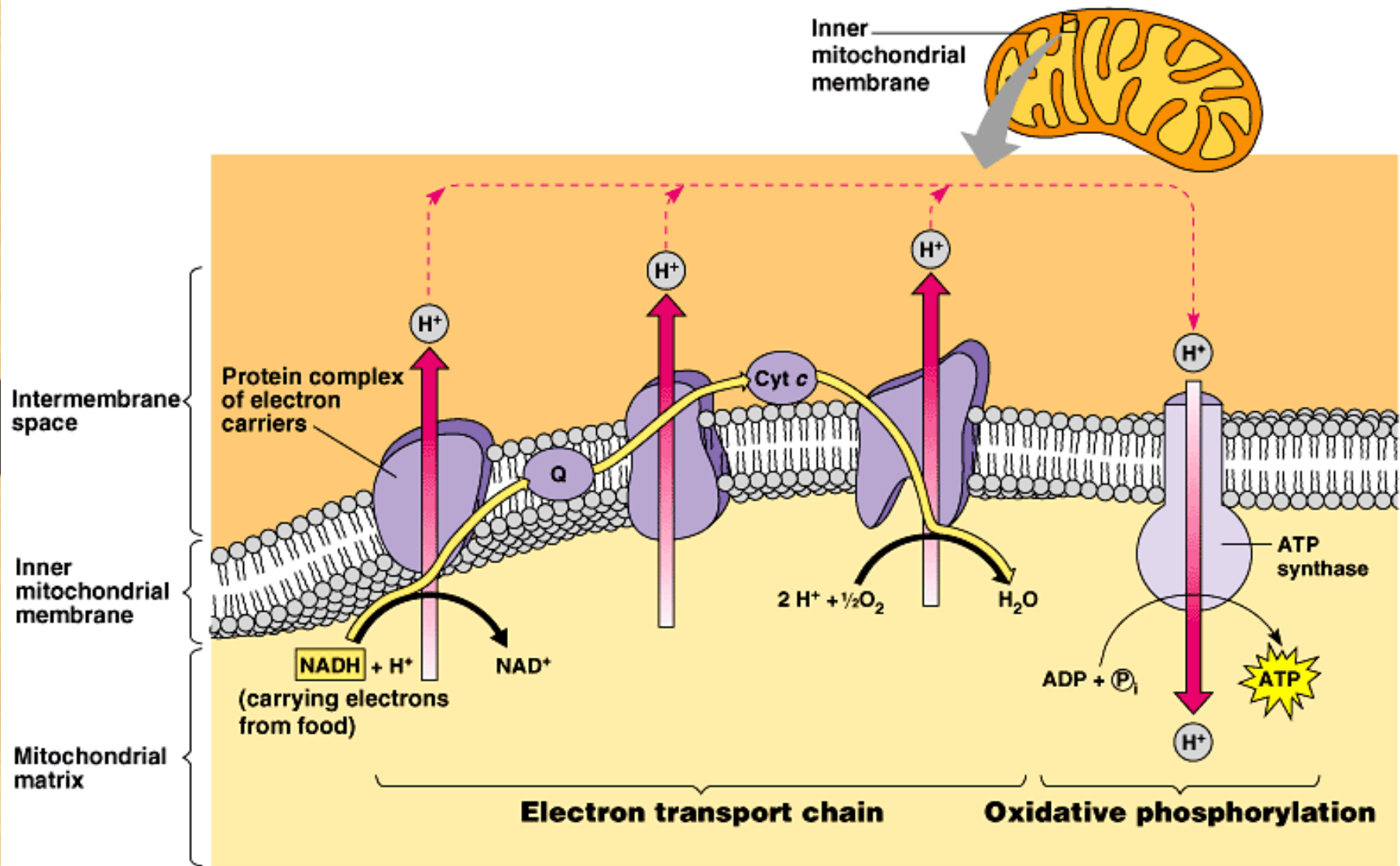
**KREBS
CYCLE**



STEP 4 – Electron Transport Chain (ETC)

- Electron carrier molecules
- membrane proteins in the cristae
- Shuttles electrons that release energy used to make ATP through Chemiosmosis
- Sequence of reactions that prevents energy being released in 1 explosive step





Electron Transport Chain

- Together, glycolysis and Krebs Cycle have produced only a net gain of 4 ATP molecules
- NADH & FADH₂ which gained electrons in these processes will release their electrons in the ETC to form the rest of the ~36 ATP made from one glucose.

Electron Transport Chain

- NADH & FADH₂ will donate their electrons to the system of electron carrier molecules embedded in the inner membrane of the mitochondrial membrane.
 - **Remember:** cristae increase surface area for chemical reactions to occur
- Most of the ETC is composed of various proteins.
 - Their job is to remove protons (H⁺) from NADH and FADH₂ to create an electrochemical gradient between the intermembrane space and the mitochondrial matrix.

Electron Transport Chain

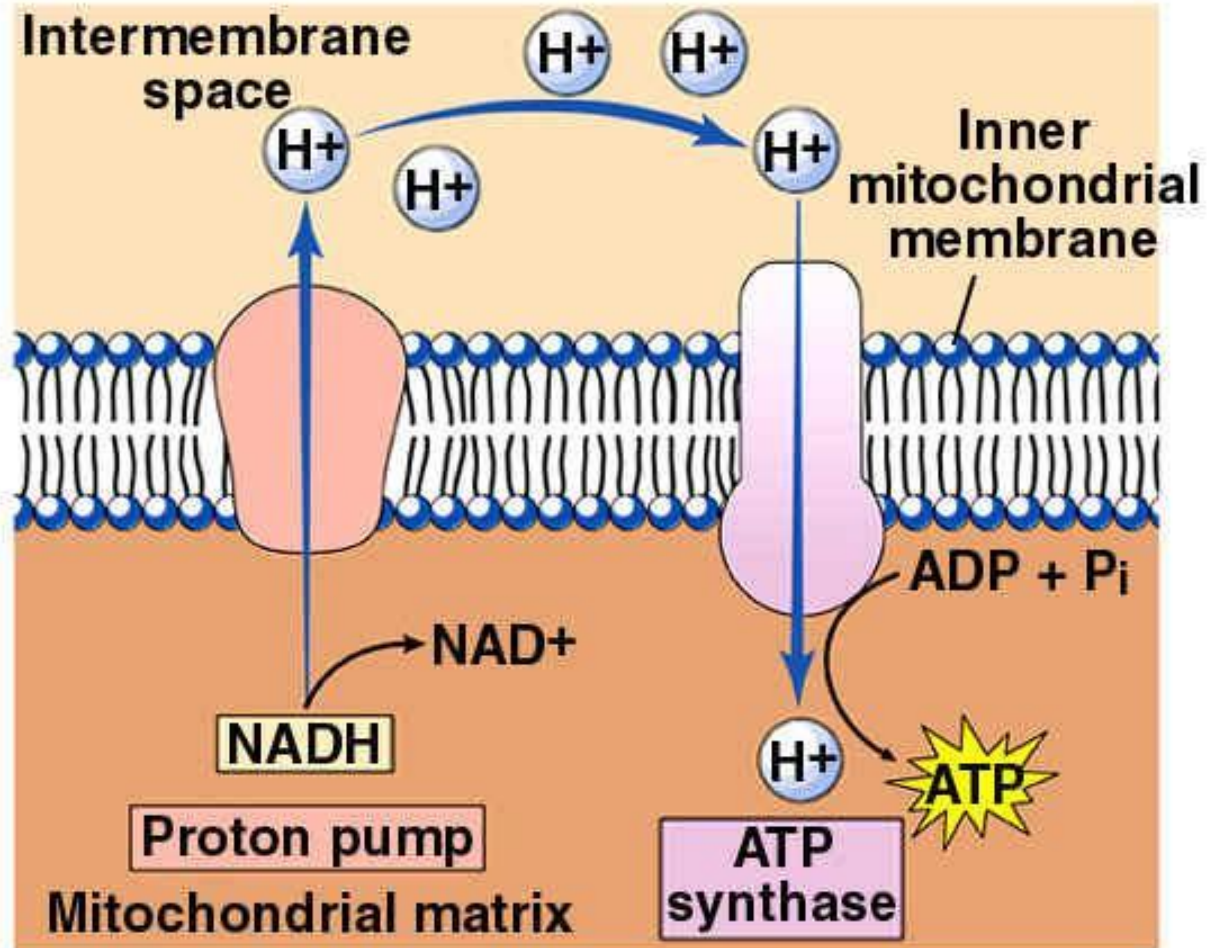
- Using energy from the exergonic electron chain flow, the ETC creates a proton gradient by **pumping H⁺ from the matrix to the intermembrane space**
- This proton gradient is maintained because the membrane's bilayer is impermeable to H⁺, thus preventing diffusion back to the matrix

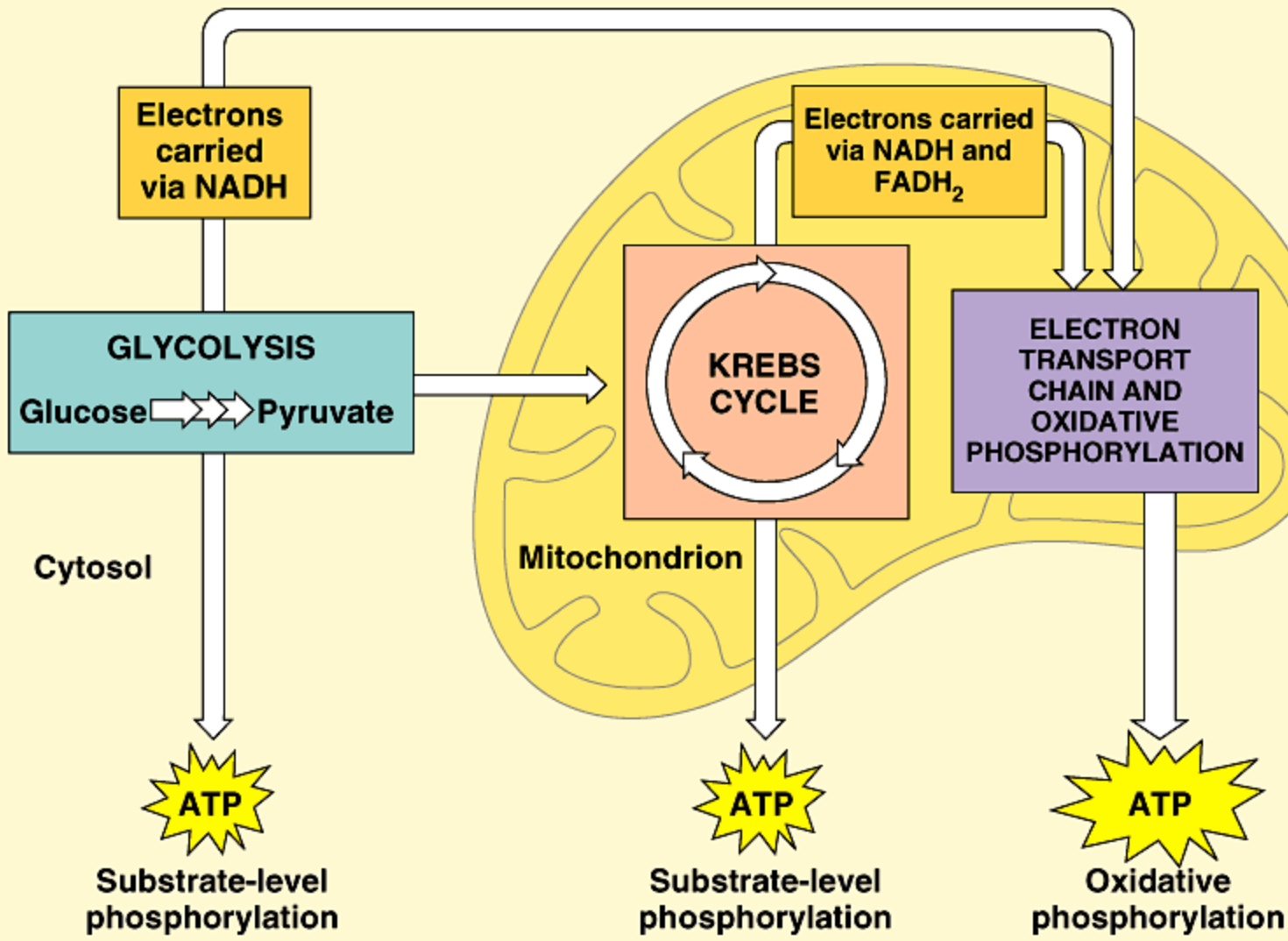
ETC - Chemiosmosis

- ATP synthase uses the potential energy stored in a proton gradient to make ATP by allowing H^+ to diffuse down the gradient, back across the membrane
 - This provides the power to allow the oxidative phosphorylation of ADP and P_i into ATP.

Copyright © The McGraw-Hill Companies, Inc. Permission required for reproduction or display.

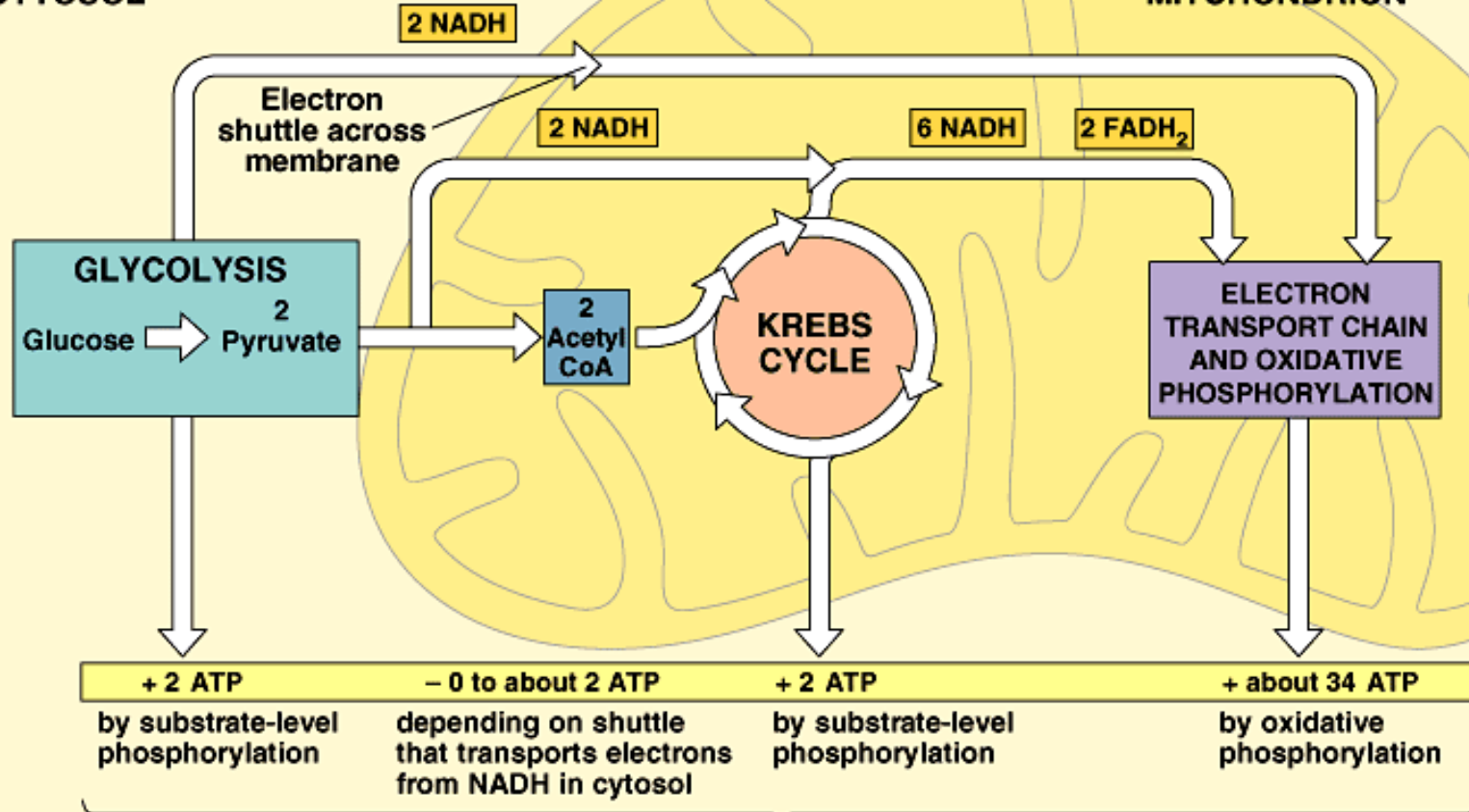
Chemiosmosis





CYTOSOL

MITCHONDRION



Maximum per glucose:

About 38 ATP

Cellular Respiration

mitochondria | aerobic

(2) pyruvate



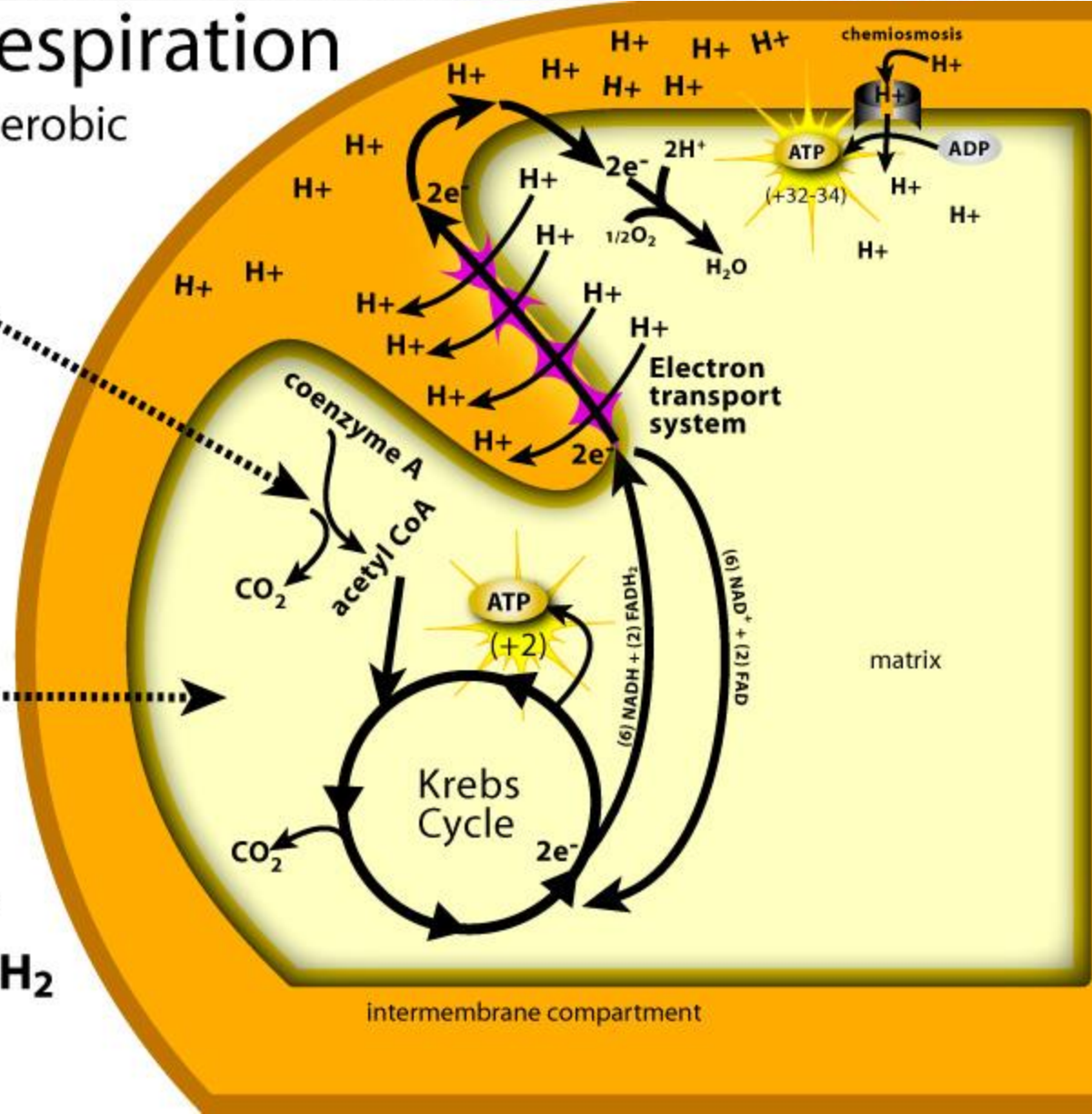
cytoplasm

(2)

NADH

Net: 34-36 ATP

6 NADH, 2 FADH₂



intermembrane compartment

Versatility of Catabolic Reactions

- Cell respiration can utilize other molecules from food to start the process
 - Proteins are broken into amino acids with NH_3 removed, which feeds into glycolysis or Krebs
 - Carbs such as starch is broken down to glucose, which feeds into glycolysis
 - Fats are broken into glycerol and fatty acids
 - Glycerol is changed to G3P (simple sugar)
 - Fatty acids are broken into acetyl CoA

Cell Respiration Regulation

- Negative feedback mechanism
- One of the enzymes in glycolysis is an allosteric enzyme
 - Products made will bind to the enzyme and inhibit the activity