Diffusion Pre-Lab Questions

Data Chart

	Initial	Solution Color		Presence of Glucose	
	Contents	Initial	Final	Initial	Final
Bag	H ₂ O & Corn Starch				
Beaker	H ₂ O & IKI				

If glucose was added to the bag, what would be expected to occur if you learned that glucose is permeable to the plastic bag?

Analysis

- 1. Which substance(s) are entering the bag and which are leaving the bag? What experimental evidence supports your answer?
- 2. Explain the results you obtained. Include the concentration differences and membrane pore size in your discussion.
- 3. Quantitative data uses numbers to measure observed changes. How could this experiment be modified so that quantitative data could be collected to show that water diffused into the dialysis bag?
- 4. Based on your observations, rank the following by relative size, beginning with the smallest: glucose molecules, water molecules, IKI molecules (Iodine potassium-iodide), membrane pores, starch molecules.
- 5. What results would you expect if the experiment started with a glucose and IKI solution inside the bag and only starch and water outside? Why?
- 6. What is water potential? What are the two equations associated with water potential?
- 7. Draw the experimental setup for Part A of the lab in the space provided. Complete the following chart as it relates to the experimental setup

	Dialysis Tubing Bag	Water Solution
High / Low Solute		
concentration		
High / Low Water		
concentration		
Hypertonic / Hypotonic		
High / Low Water		
potential		
Direction of water		
movement		

Hypothesis: Write a statement to describe the relationship between the amount of sucrose in the bag and the change in mass when placed in water.